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STATIC FATIGUE BEHAVIOR OF STRUCTURAL CERAMICS IN A CORROSIVE ENVIRONMENT

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ABSTRACT

Flexure testing was used to determine the effects of sodium sulfate induced corrosion on the static fatigue behavior of several structural ceramics between 800°C and 1200°C. The results showed that the static fatigue behavior of a high purity, fully-dense alumina and a Ce-TZP are unaffected by this corrosive environment. However, the static fatigue behavior of a MgO-doped Si₃N₄ and, to a lesser degree, a Y-TZP are affected by the introduction of sodium sulfate.

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INTRODUCTION

A variety of structural ceramics are being considered for application in advanced heat engines. In some applications, these materials will be exposed to molten sodium sulfate (Na₂SO₄) which condenses on engine components when ingested NaCl reacts with sulfur impurities in the fuel. How long-term exposure to these corrosive conditions affects the static fatigue behavior of the ceramics at elevated temperatures is of critical importance to their successful application in the next generation of advanced engines.

The actual mechanisms of Na₂SO₄-induced corrosion in TZPs¹⁻⁵ and Si₃N₄⁶⁻¹² have been studied. It has been shown that the hot corrosion resistance of TZP materials is dependent on the stabilizers rather than the zirconia.⁴ Stabilizers are added to zirconia in order to retain the tetragonal phase in a metastable state at room temperature. This allows for the production of a zirconia material, which is essentially 100% tetragonal, having an unusual combination of high strength and toughness. Of the two most common TZP stabilizers, CeO2 and Y₂O₃, the thermodynamic data indicates that CeO₂ is much more difficult to sulfate than Y₂O₃. The sulfation of these rare earth oxides effectively leaches the oxide from the TZP causing surface destabilization and a reduction in properties. Thus, CeO2-stabilized zirconia will be more resistant to corrosion by Na₂SO₄ than the Y₂O₃-stabilized zirconia. Previous work by the authors 13 shows that after 500 hours at 1000°C in the presence of Na₂SO₄, the room temperature strength of a Y₂O₃-stabilized zirconia (Y-TZP) decreases by ≈30% when compared to the strength in the as-received condition. This reduction is ≈10% greater than the loss seen due to overaging alone. On the other hand, the room temperature strength of the CeO₂-stabilized zirconia (Ce-TZP) is unaffected by the same treatment. This supports the thermodynamic data.

Silicon-based ceramics such as Si_3N_4 rely on a thin surface layer of SiO_2 for protection against oxidation. At temperatures above its melting point, Na_2SO_4 dissociates into Na_2O and SO_3 . The Na_2O that is formed reacts with the SiO_2 protective layer to form a sodium-silicate glass which allows for extensive corrosion and degradation of the ceramics.¹¹

Although the corrosion mechanisms have been studied, there has been no work performed, to the authors knowledge, which examines the mechanical properties of structural ceramics at elevated temperatures in a corrosive environment. This report summarizes an effort to determine the effects of hot corrosion on the mechanical properties of several structural ceramics between 800°C and 1200°C.

EXPERIMENTAL PROCEDURE

Four commercially available ceramics, a yttria-TZP, a ceria-TZP, an alumina, and a Si_3N_4 , (see Table 1) were obtained and machined into flexure specimens of the following dimensions: 1.5 mm x 2 mm x 25 mm. The specimens were carefully ground by a surface grinder such that the surface striations were parallel to the long axis. All four long edges were chamfered $\approx 45^{\circ}$ to a depth of $\approx 0.15 \ \mu m$; the specimens were machined according to Reference 14.

Table 1. LIST OF MATERIALS TESTED

Material	Manufacturer	Code	Process	Additives	Grain Size (µm)
Y-TZP	NGK Insulators	Z-191	Sintered	Y ₂ O ₃	0.3
Ce-TZP	Ceramatec	CZ203	Sintered	CeO ₂ /Al ₂ O ₃	1.2
Si3N4	Norton	NC-132	Hot Pressed	MgO	3-6
Al ₂ O ₃	Coors Ceramics	AD-999	Sintered	None	3-6

Stepped-temperature stress-rupture (STSR) testing was done following the procedure outlined by Quinn and Katz. 15 Tests were done on as-received specimens and on specimens with 10 to 20 mg/cm² of Na₂SO₄. STSR testing allows rapid screening of a material's static fatigue behavior over a wide range of temperatures and stresses using a small number of specimens. The procedure involves loading a bend bar onto a four-point flexure fixture (inner and outer spans of 10 mm and 20 mm, respectively) that is contained in a furnace. The furnace is then heated to 800°C in two hours, in air, with no stress applied to the specimen. Upon reaching this temperature, a predetermined stress is applied. Should the specimen survive 24 hours at this temperature, the furnace is then heated (in ≈10 minutes) to 900°C, and again the specimen is allowed to soak for 24 hours while under the same stress. This cycle is repeated for 1000°C, 1100°C, and 1200°C. If the specimen fractures or excessive creep occurs, the power to the furnace is shut off by a microswitch. The time of fracture is then denoted on the STSR plot using an arrow with the applied stress that caused fracture above the arrow. The symbols for the STSR plot are (\leftarrow) failure occurred upon application of the stress at 800°C, (→) survived the full test cycle through 1200°C, and (↓) denotes the time of failure between loading but before the full cycle is complete.

Long duration stress rupture (SR) tests were conducted on specimens in the as-received condition and with 10 to 20 mg/cm² Na₂SO₄ added. Tests were carried out following the procedure outlined for the STSR tests, but instead of varying the temperature it was kept constant at 1000°C. In addition, the tests were allowed to run until the specimen either failed or 500 hours had elapsed.

Specimens which survived 500 hours without fracture were then broken in four-point flexure at room temperature to determine the retained strength. Special care was taken to ensure that the surface placed in tension during SR testing was also in tension during this test, and that the upper and lower spans were aligned in the same location as during the SR test.

RESULTS AND DISCUSSION

Stepped-Temperature Stress-Rupture

STSR tests for all four ceramic materials are summarized in Figures 1 through 4. Tests on the Y-TZP material (see Figure 1) show that the addition of Na₂SO₄ has an adverse effect on the material's high temperature static fatigue behavior. This effect is pronounced at the 900°C and 1000°C levels under stresses of 200 to 300 MPa. The material can no longer sustain stresses greater than 200 MPa through these temperature levels. This shift is probably due to the depletion of the Y₂O₃ from the surface which allows the tetragonal-to-monoclinic transformation of the zirconia to occur.^{2,3}

STEPPED-TEMPERATURE STRESS-RUPTURE

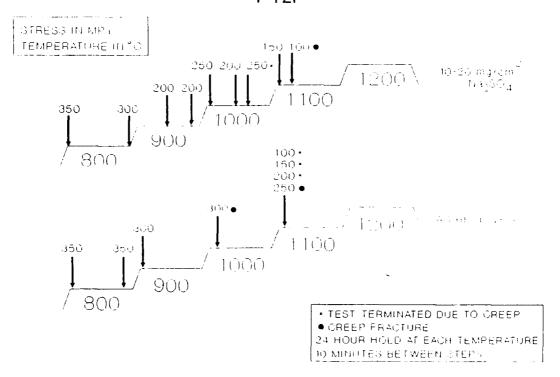


Figure 1. STSR results for Y-TZP with and without sodium sulfate.

Sodium sulfate melts at $\approx 884^{\circ}$ C; thus one would expect little, if any, change in the static fatigue behavior at 800° C and, indeed, none is encountered. Since the tetragonal zirconia is stable at $\approx 1170^{\circ}$ C, the depletion of the Y_2O_3 from the surface at or near this temperature will not initiate the transformation to monoclinic zirconia. Therefore, at stresses below 200 MPa the static fatigue behavior appears unaffected.

The results of the STSR tests of the Ce-TZP and alumina (see Figures 2 and 3) indicate that the high temperature static fatigue behavior of these materials is unaffected by the addition of Na₂SO₄. Based on the room temperature data,¹³ the thermodynamic data for the Ce-TZP,²⁻⁴ and a study¹⁶ showing that high purity, fully-dense alumina is very resistant to hot corrosion by a similar molten alkali salt, these results are not surprising. However, the retention of the high temperature static fatigue behavior of the Si₃N₄ (see Figure 4) was not expected based on the room temperature strength degradation seen after thermal exposure in the presence of Na₂SO₄.¹³ There are several possible explanations for this behavior:

- 1. The STSR test is not sensitive enough to allow the effects of corrosion to be seen.
- 2. During the first step of the SISR test, there is sufficient time for a protective layer to form, possibly through oxidation or the formation of a sodium-silicate glass, which prevents corrosion.
 - 3. The addition of the Na₂SO₄ does not effect the static fatigue behavior in this temperature range.

STEPPED-TEMPERATURE STRESS-RUPTURE

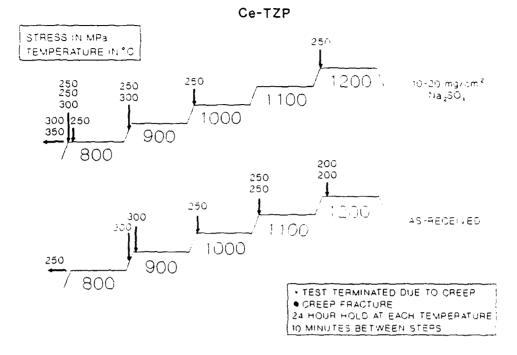


Figure 2. STSR results for Ce-TZP with and without sodium sulfate.

STEPPED-TEMPERATURE STRESS-RUPTURE

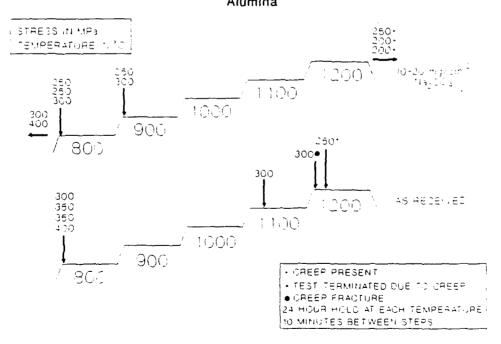


Figure 3. STSR results for alumina with and without sodium sulfate.

STEPPED-TEMPERATURE STRESS-RUPTURE

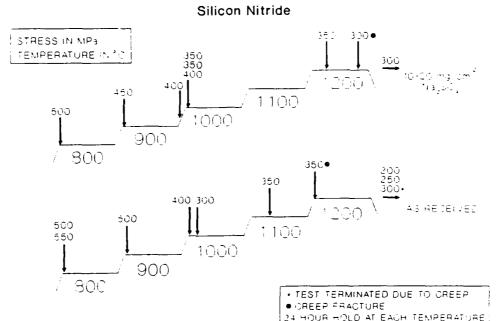


Figure 4. STSR results for silicon nitride (NC-132) with and without sodium sulfate.

10 MINUTES BETWEEN STERS

In support of 3 Rowcliffe and Huber¹⁷ examined the effect of hot gas corrosion on silicon nitride by combusting and passing fuel-containing sulphur and sodium over notched specimens at 900°C and 950°C, respectively. They found that the sodium addition to the fuel did not noticeably effect the failure time at these temperatures. However, our room temperature results lead us to believe that 1 and/or 2 are more likely. Stress rupture testing, which will be discussed in the next section, clearly shows that 1 is the obvious explanation for this surprising behavior.

Stress Rupture Tests

Long duration stress rupture (SR) tests were done on only two of the four ceramics. The two ceramics were the Ce-TZP and Si₃N₄ materials. The Ce-TZP was selected because plasma-sprayed CeO₂-ZrO₂ has been patented as a protective coating for superalloys exposed to vanadium and SO_x impurities in gas turbine engines. Silicon nitride was selected because this material is being considered and, in certain applications, incorporated as monolithic components in advanced engines.

Ce-TZP

Stress rupture testing of the Ce-TZP is shown in Figure 5. There is a significant amount of scatter in the data both with and without Na₂SO₄. The amount of scatter seen in Figure 5 is common to TZP materials. Due to this scatter, no specific relationship between stress and time-to-failure at 1000°C could be determined. However, some general conclusions can be drawn. First, at a stress in excess of 250 MPa failure occurs on loading, while at 200 MPa or less the specimen can survive for 500+ hours without failure. Secondly, the static fatigue behavior does not change when Na₂SO₄ is added. Based on the thermodynamic data²⁻⁴ and the authors previous work¹³ the latter was not unexpected.

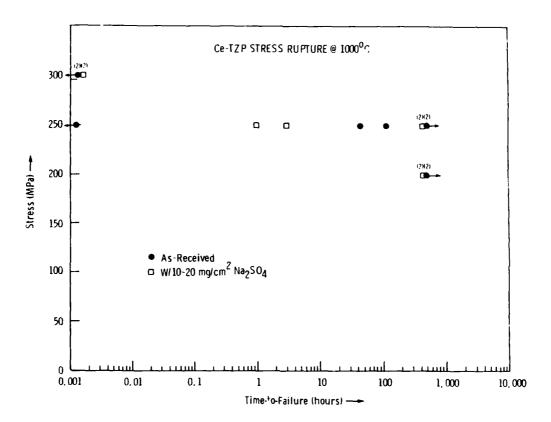


Figure 5. Stress rupture results for Ce-TZP at 1000°C. The numbers in parenthesis indicate the number of specimens which failed at that time under the applied stress.

Si₃N₄ Without Sodium Sulfate

There is extensive information on the static fatigue behavior of MgO-doped Si₃N₄ in flexure. By superimposing the data from this study onto that from Quinn²⁵ (see Figure 6), it is obvious that there is excellent agreement between the data even though there is a difference in the

size of the specimen cross sections. The slight differences in behavior is attributed to billet-to-billet variations and to subtle deviations in the machining of the specimens at the high stresses.

At stresses of 500 MPa or greater, optical fractography showed what appears to be a fast fracture-type failure (see Figure 7). Scanning electron microscopy (SEM) fractography revealed that machining damage was the cause of failure. Specimens which were subjected to a 450 MPa stress began to show evidence of slow crack growth (SCG); this was especially obvious in the specimen which failed after ≈56 hours (see Figure 8). The other specimens which failed at 15 and 20 hours, respectively, showed signs which may be interpreted as SCG, but more in-depth fractography is needed to determine if SCG is present. Specimens which survived 500+ hours under a 400 MPa applied stress without failure showed no signs of creep.

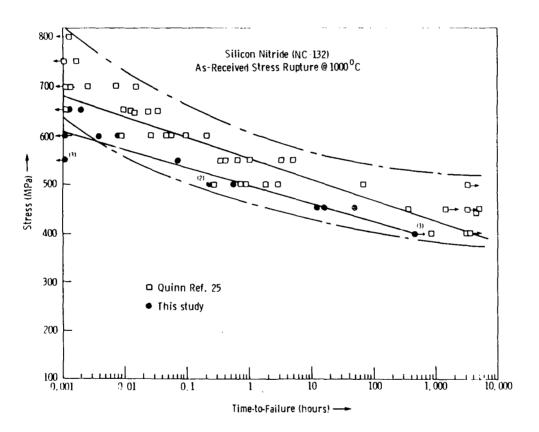


Figure 6. Comparison of stress rupture results at 1000°C for silicon nitride (NC-132). The dashed lines indicate the approximate performance boundaries. The line for the data from Reference 25 is drawn using linear regression. The line for the data from this study was drawn to show the trend of the data.

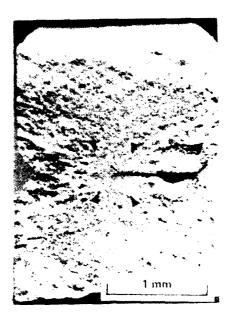


Figure 7. Optical photograph of an as-received fracture surface of NC-132 stress rupture specimen showing the fast fracture-type failure. Specimen failed after ~0.4 hours under 500 MPa stress.

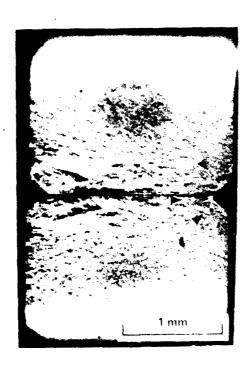


Figure 8. Optical photograph of NC-132 which failed after =56 hours under 450 MPa stress showing the slow crack growth (SCG) region.

Si₃N₄ With Sodium Sulfate

The introduction of 10 to 20 mg/cm² of Na₂SO₄ to the SR test greatly reduces the static fatigue behavior of this Si₃N₄ at 1000°C, as seen in Figure 9. This is contrary to our STSR test results. The reason for this difference is the time frame of each test. The SR test accentuates minor changes in the time-to-failure at a specific temperature, while the STSR test is for screening purposes and will only accentuate deficiencies that reduce the time-to-failure by several orders of magnitude. The later is indicative of what happened to the Y-TZP material in this study, as shown in Figure 1.

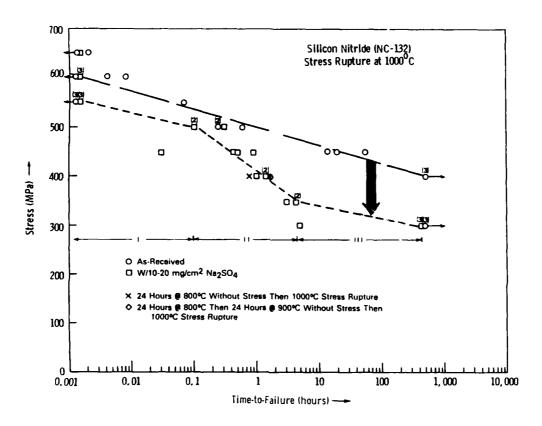


Figure 9. Stress rupture results at 1000°C for NC-132 with and without sodium sulfate. Numbers in parenthesis indicate the number of specimens which failed at that time under the applied stress.

In order to determine if a protective layer was formed during the 800°C or 900°C steps of STSR testing, several tests were run to duplicate these conditions before long-time stress rupture was done. Two specimens were heated to 800°C and held for 24 hours without applying a stress. After 24 hours, the specimen was heated to 1000°C, a 400 MPa stress was applied, and the specimen was allowed to .un until failure or 500 hours had accumulated; these tests are designated by the x in Figure 9. A third test (designated by the \diamond in Figure 9) was identical except that an additional 24 hour hold at 900°C without the applied stress was added before stress rupture at 1000°C. In all three instances, the time-to-failure was virtually the same as the regular SR tests indicating that no protective layer was being formed.

The addition of 10 to 20 mg/cm² of Na₂SO₄ made fractography extremely difficult. Upon failure, a glassy phase, molten Na₂SO₄ and/or corrosion product, flowed over the fracture surface obscuring the failure marks (see Figure 10). A mild HF solution was used to remove this glassy phase, but, in doing so, it also removed the glassy phase inherent to the material that highlights the details on the fracture surface.



Figure 10. Optical photograph showing that the glassy phase and/or corrosion product flowed over the fracture surface after failure in a stress rupture test.

This addition of Na₂SO₄ also yields a SR profile that has three regions of interest. In Region I, the slope of the line is essentially the same as in the as-received condition indicating that the failure mechanism is the same in both cases, but the time-to-failure is reduced slightly. This may be due to a subtle increase in the size of the machining damage by the corrosive environment. This enhancement is not discernible when the fracture surfaces are examined.

The slope of the profile in Region II changes significantly due to a change in the strength-limiting defect. Fractographic analysis shows that corrosion pits ≈ 10 to 15 μ m in size are being formed (see Figure 11). It is the formation of these pits rather than slow crack growth that reduces the time-to-failure in Region II.

At stresses below 350 MPa, Ragion III, the slope of the line is the same as in the asreceived condition, but the time-to-failure is reduced. This indicates that the corrosion pits which form reach a certain size and then stop growing. This is a definite possibility since the sodium sulfate is not replenished during the test. Thus, once the sodium sulfate is depleted, the pit will stop growing. As a result, when stresses lower than 350 MPa are applied, the specimen can survive much longer than when stresses above 350 MPa are applied.



Figure 11. Corrosion pit (=10 to 15 µm) which is formed when sodium sulfate is added to the stress rupture tests.

The size and location of Regions II and III may change if the time of the test was significantly increased beyond 500 hours, or if the Na₂SO₄ was replenished during the test.

Retained Strength at Room Temperature

The room temperature-retained strength of the Ce-TZP and Si₃N₄ specimens which survived 500+ hours at 1000°C during stress rupture testing is summarized in Table 2. The retained strength could not be compared to the strength after exposure from Reference 13 because different specimen sizes were used in each test.

Table 2. RETAINED STRENGTH

Material/Condition	SR Stress (MPa)	Retained Strength (MPa)	
Ce-TZP:			
As-Received	200 250	Average = 490 Average = 512	
w/10-20 mg/cm ² Na ₂ SO ₄	200 250	Average = 477 Average = 408	
Silicon Nitride:			
As-Received	300 Average = 906 400 Average = 640		
w/10-20 mg/cm ² Na ₂ SO ₄	300 400	A je = 725 No specimens survived	

As expected, the strength of the Ce-TZP did not change after 500 hours under an applied stress with or without sodium sulfate.

The retained strength of the Si₃N₄ did change, not only with the addition of the Na₂SO₄, but also with an increase in applied stress. The change after the addition of Na₂SO₄ is due to the formation of corrosion pits as previously seen in this study (see Figure 11) and in Reference 13. The change with applied stress is due to slow crack growth. Figures 12a and 12b are the fracture surface of the as-received specimen that were subjected to 400 MPa and 300 MPa stress, respectively. In Figure 12a, a SCG zone is evident but none is evident in Figure 12b. This SCG zone increases the flaw size thus decreasing the retained strength. If the SR tests were to run for 1000 hours instead of 500 hours, the specimens subjected to the 400 MPa stress may have failed, and, in those subjected to a 300 MPa stress, a visible SCG zone may have started to form. No SCG was observed on the retained strength fracture surfaces of the Si₃N₄ after SR with the sodium sulfate present.

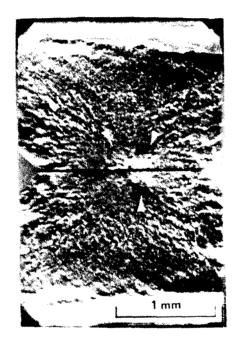


Figure 12a. Optical photograph of the fracture surface of an as-received NC-132 specimen which was used to determine the retained strength after surviving 500+ hours at 1000°C under a 400 MPa stress. SCG region is outlined.

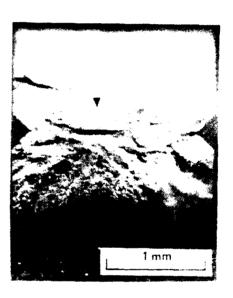


Figure 12b. Optical photograph of the fracture surface of a NC-132 specimen which was used to determine the retained strength after surviving 500+ hours at 1000°C under a 300 MPa stress. No SCG region is seen.

CONCLUSIONS

This study showed that the static fatigue behavior of a high purity, fully-dense alumina and a Ce-TZP are unaffected by the addition of 10 to 20 mg/cm² Na₂SO₄. On the other hand, a Y-TZP and a MgO-doped Si₃N₄ are affected by anis addition. STSR tests showed that the Y-TZP is affected between 900°C and 1000°C under stresses of 200 to 300 MPa. This is probably due to the depletion of the Y₂O₃. Initial STSR testing of the Si₃N₄ showed no effect, but SR tests did show degradation in the static fatigue behavior. This discrepancy is due to the time frame of each test. The SR tests showed that at stresses above ≈ 500 MPa.

the static fatigue behavior is slightly reduced. Below 500 MPa, but above 300 MPa, the static fatigue behavior is significantly reduced. Below 300 MPa, the behavior appears unaffected.

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